

## Potensi batu bara lokal teraktivasi untuk penanganan limbah cair benzene dan toluena

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### Abstrak

Benzene and toluene are substances frequently used as solvents in chemical industries. However, these substances are also hazardous waste if they are contaminated in water. The goal of this study is to examine the effect of using simply treated coal and activated coal on the adsorption of benzene and toluene by dispersing the same amount of coals into varied concentration of benzene and toluene solution (50 unto 1000 ppm) in a batch process. Results obtained in this study include surface area of the treated coals and their effects on the adsorption capability, effects of benzene and toluene concentration on the adsorption capability, and the isotherm adsorption curves. Surface area of original coal is obtained to be 0,2687 m<sup>2</sup>/g , 4,49 m<sup>2</sup>/g for simply treated coal, and 854,2 m<sup>2</sup>/g for activated coal. The Langmuir adsorption model can adequately represent the experimental isotherm adsorption data, where the values of Langmuir constants for benzene are Q<sub>m</sub>= 138,89 mg/g and b = 45,45 on simply treated coal, and Q<sub>m</sub>=227,27 mg/g and b = 78,125 for activated coal. Meanwhile, the values of Langmuir constants for toluene are Q<sub>m</sub>= 147,06 mg/g and b = 170 on simply treated coal, and Q<sub>m</sub>=384,6 2mg/g and b = 23,23 for activated coal.

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